



SEMATECH Immersion Workshop

Photoresist Component Extraction
Measurements using
Radiochemical Analysis

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Introduction

- Objective
 - Determine the identities, amounts and rates of photoresist components extracted by water
- Method: Radiochemical analysis
 - Why?
 - Experimental design
 - The big question: What comes out?



Analysis of Radiochemical Analysis

- **Benefits**
 - Statistically very accurate
 - High signal-to-noise ratio
 - Allows a determination of absolute value
- **Detriments**
 - High overhead
 - Training
 - Requires licensing



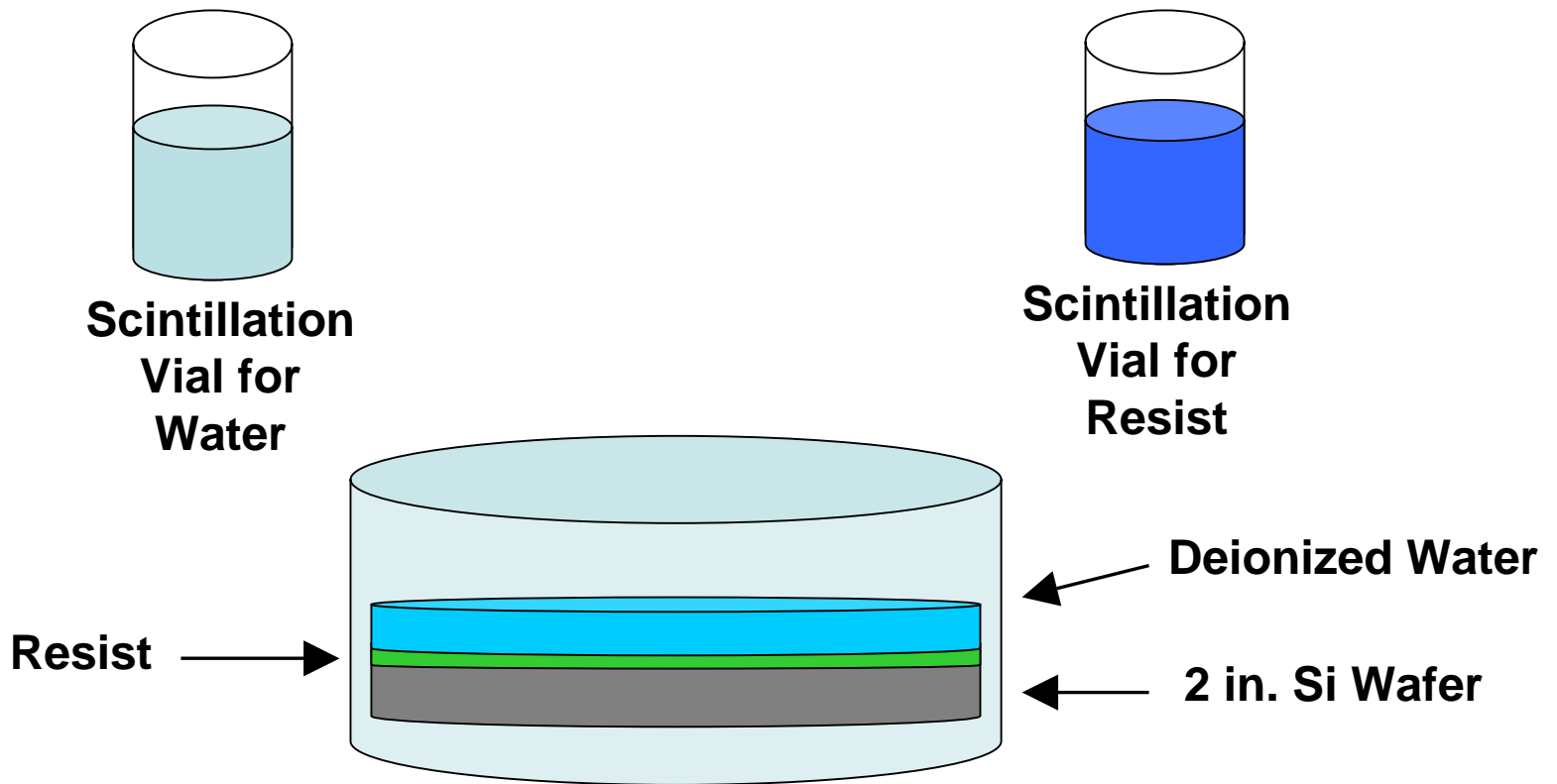
Experimental Design

- Use TOK ILP formulations
- Synthesize ^{14}C -labeled photoresist components and incorporate into formulations for extraction experiments
 - ✓ Residual casting solvent, RCS (PGMEA)
 - ✓ Photo-acid generator, PAG (TPS nonaflate)
 - Base (Triethanolamine)

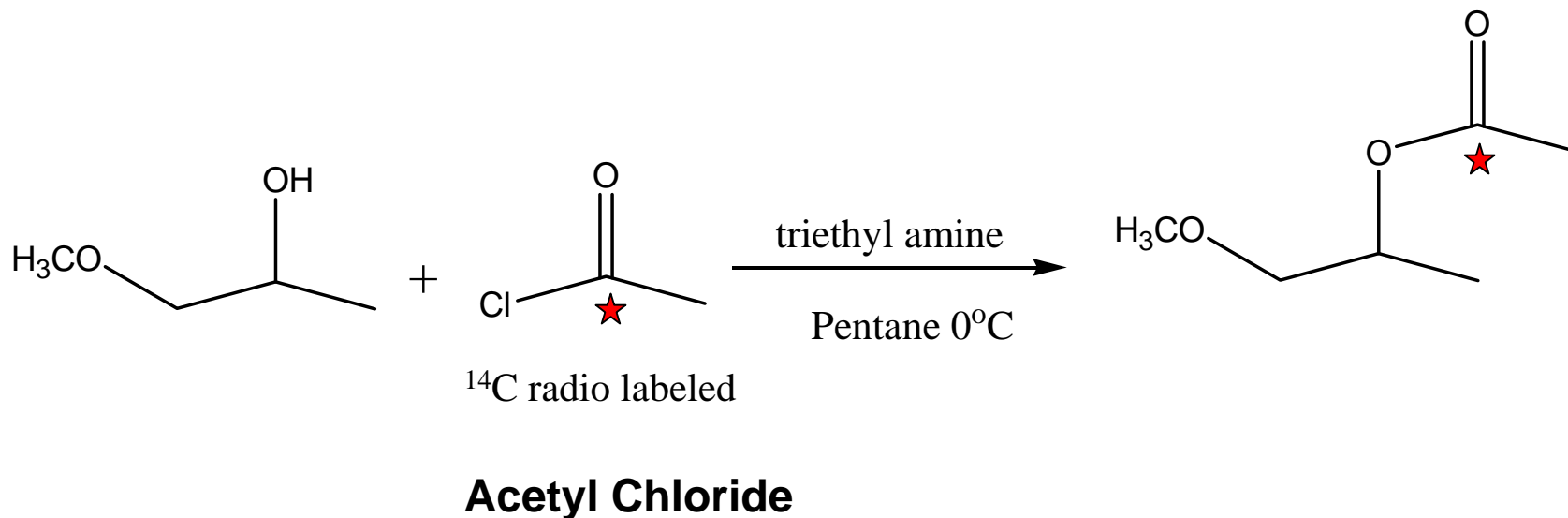


Experimental Design

- Extraction Experiments

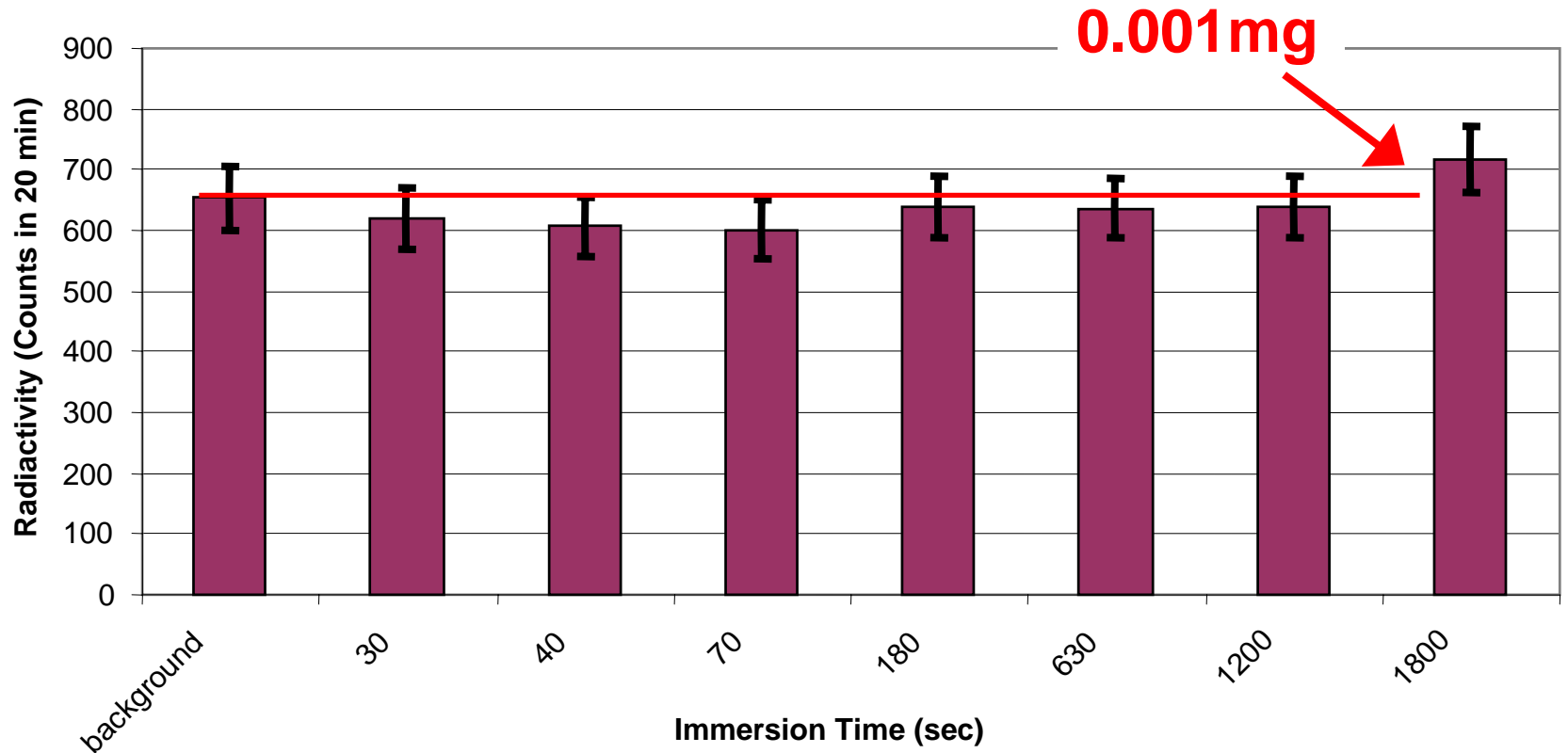


RCS Extraction Experiment - Synthesis





RSC Extraction Experiment - Results



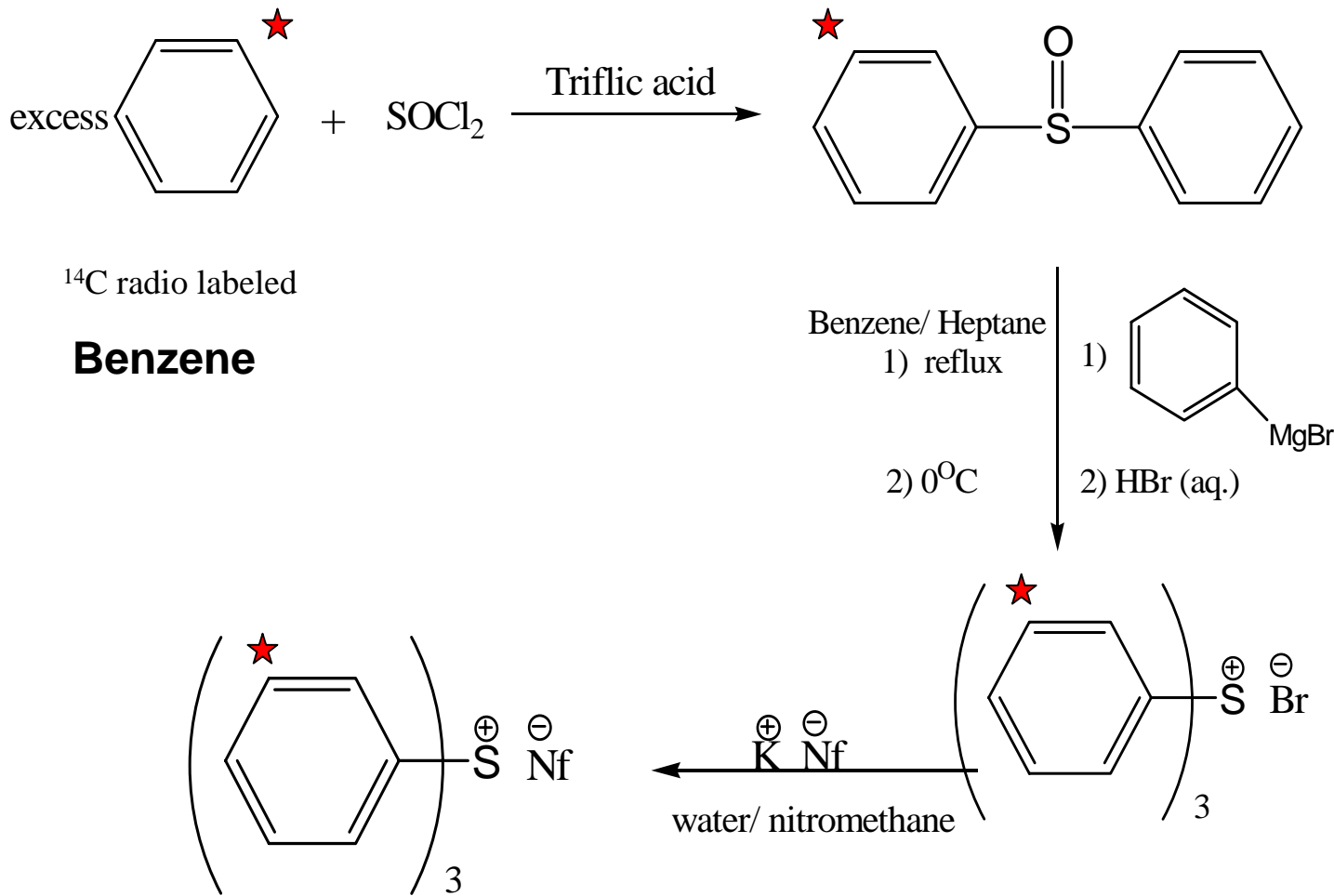
Radioactivity From ^{14}C -PGMEA in Immersion Water vs. Time



RSC Extraction Experiment - Conclusions

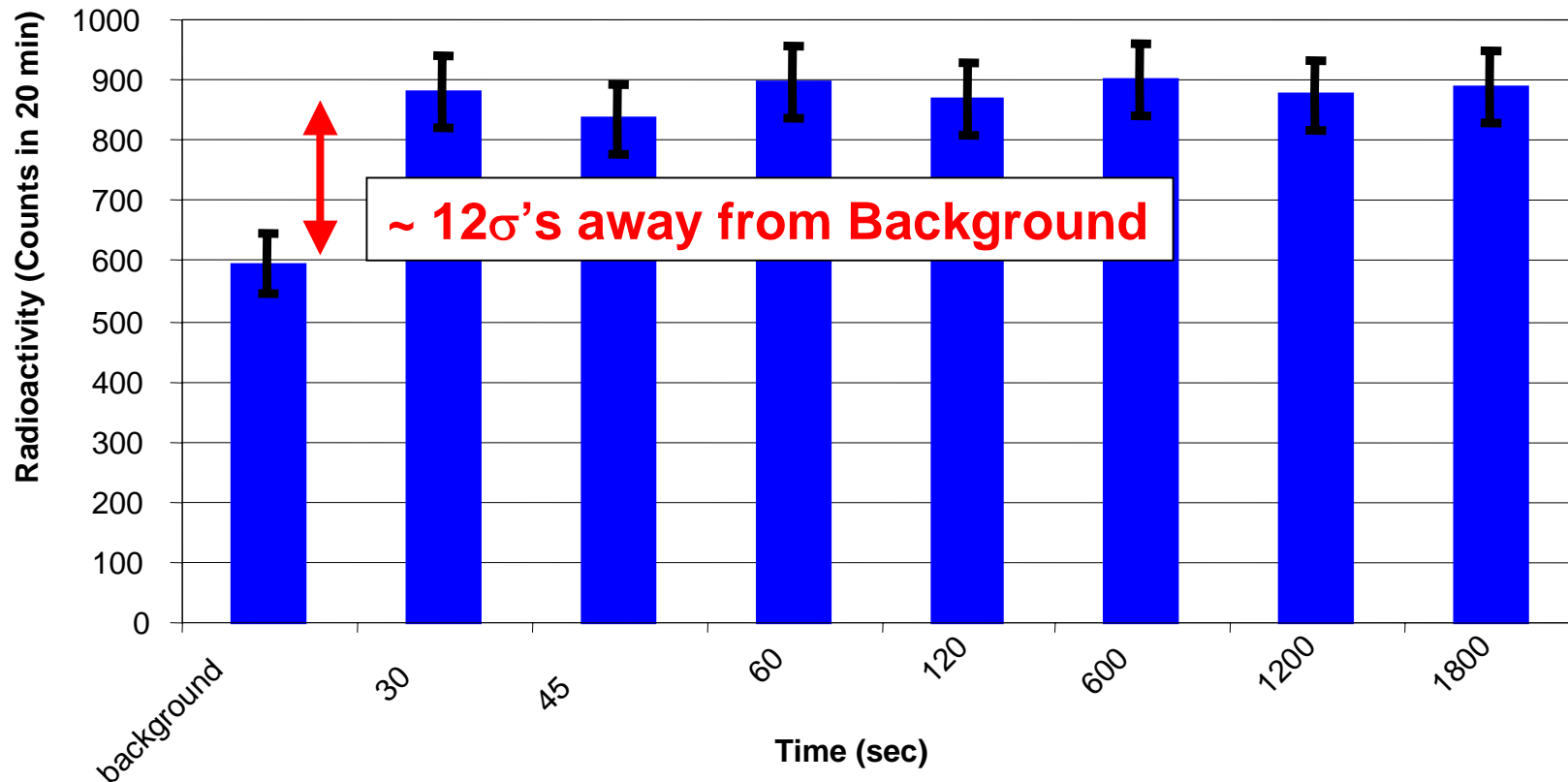
- No statistically detectable amount of RCS is extracted under 20 min.
- At most, 0.001mg of RCS is extracted after 30 min.

PAG Extraction Experiment - Synthesis





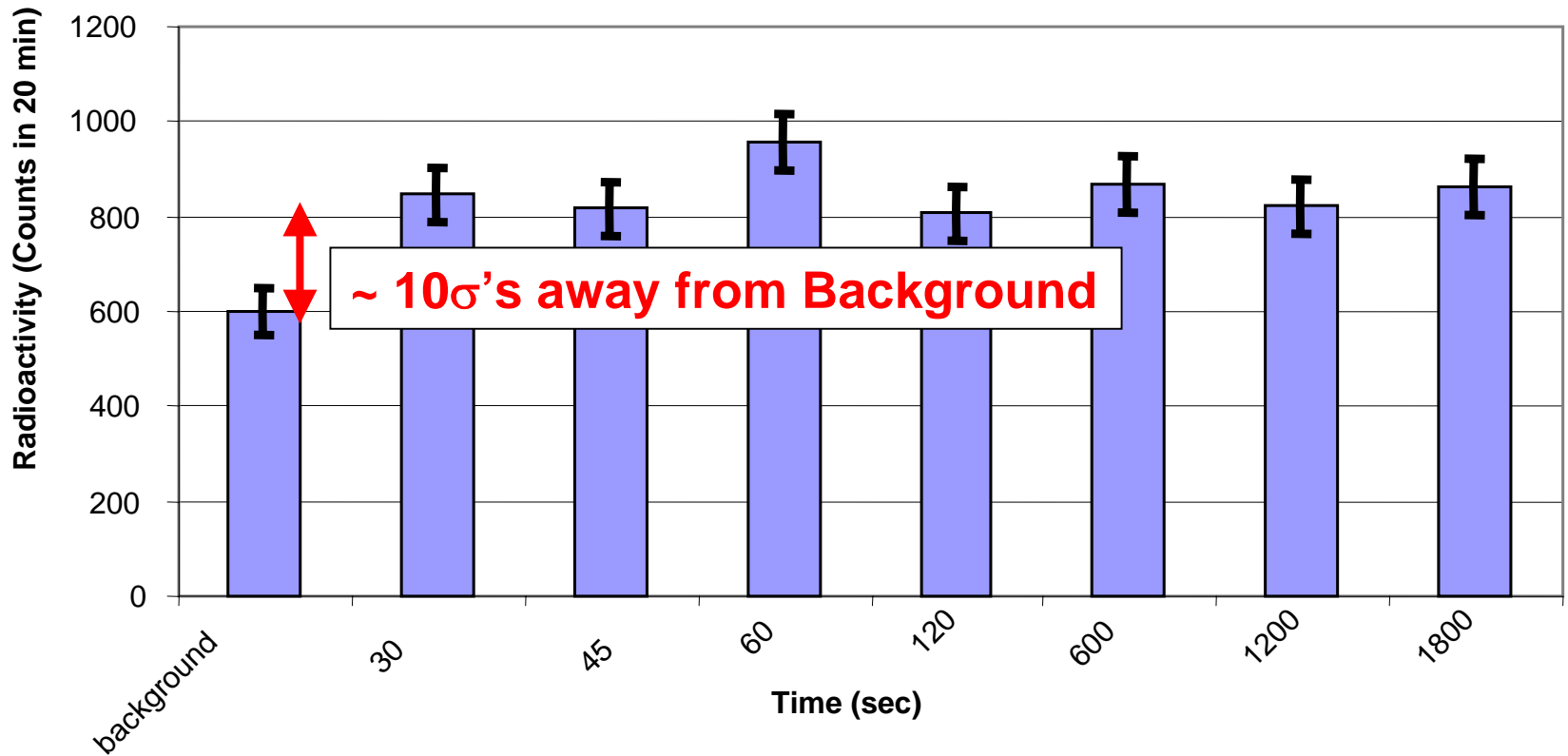
PAG Extraction Experiment - Statistical Analysis of Results



Radioactivity from Unexposed ¹⁴C-PAG in Immersion Water vs. Time



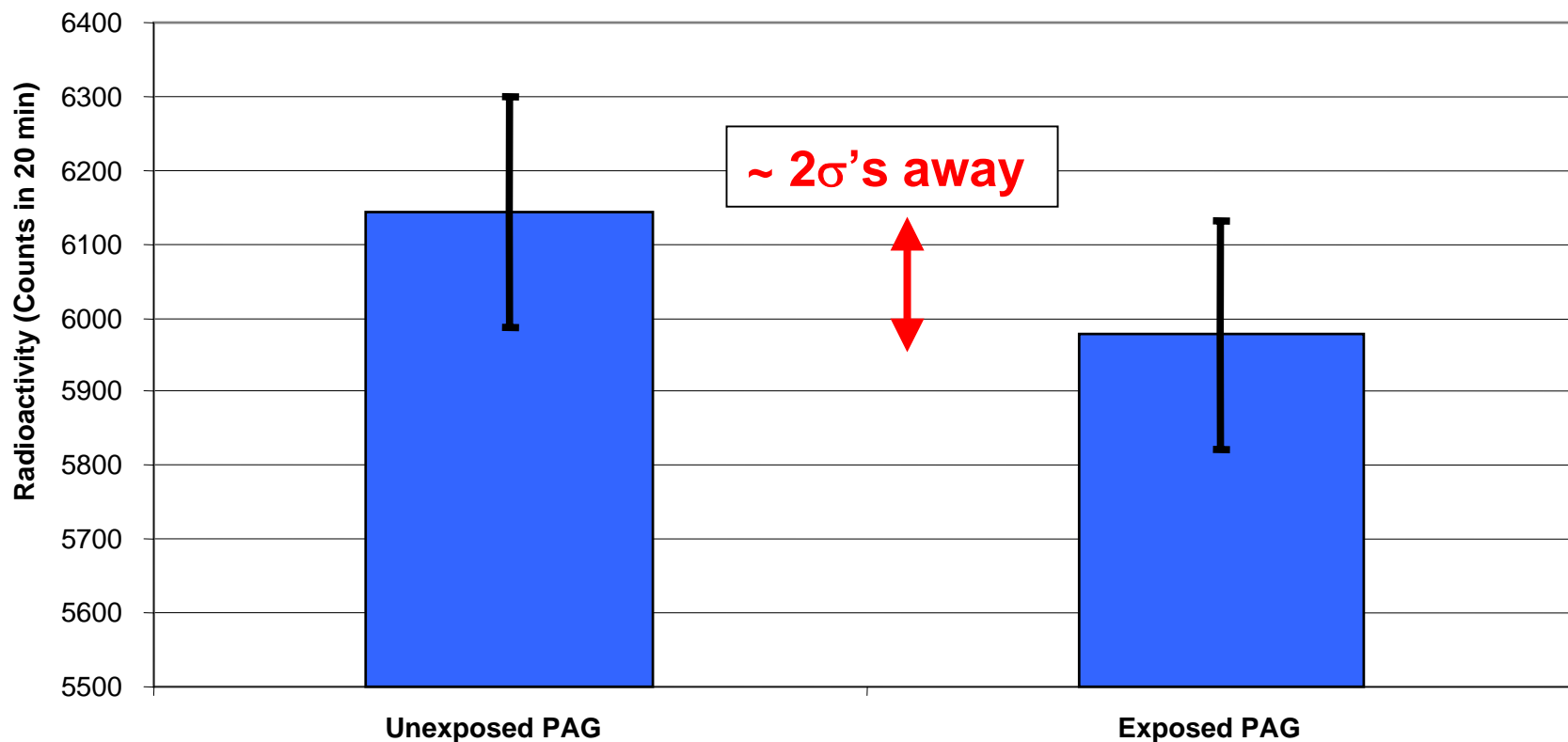
PAG Extraction Experiment - Statistical Analysis of Results



Radioactivity from Exposed ^{14}C -PAG in Immersion Water vs. Time

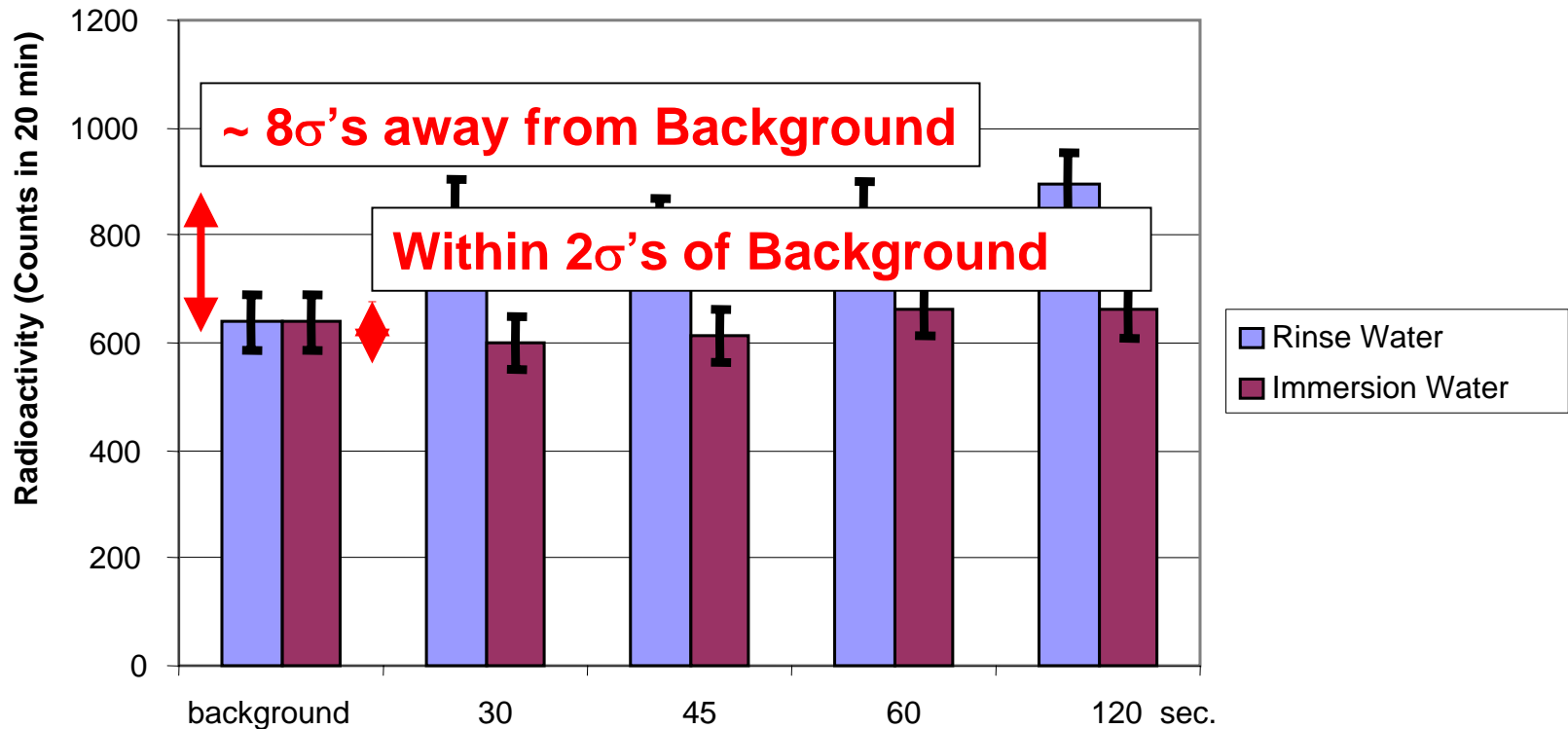


PAG Extraction Experiment - Statistical Analysis of Results



Total Radioactivity of Exposed PAG Compared to Unexposed PAG

PAG Extraction Experiment - Results of Pre-Rinse Experiment



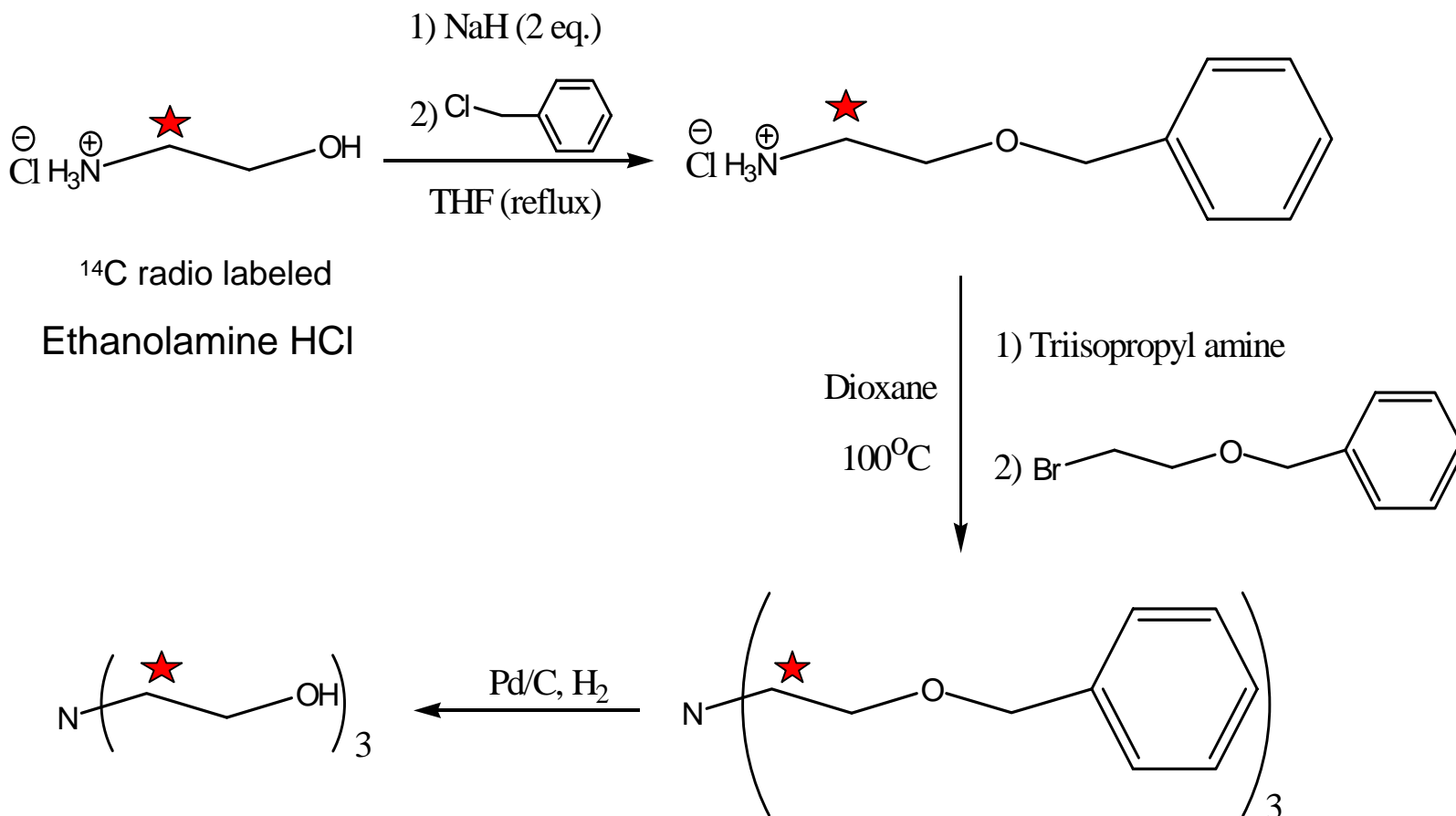
**Comparison of Amount of Radioactivity in Rinse Water to
Immersion Water**



PAG Extraction Experiment - Conclusions

- Small but detectable amount of ^{14}C -PAG is extracted under 30 sec.
- No difference between amount of exposed and unexposed PAG extracted
- No PAG extracted after 30 sec. pre-rinse

Base Extraction Experiment - Synthesis





Base Extraction Experiment - Synthesis

- Purification of base in original synthesis was extremely difficult because of triethanolamine's affinity for water
 - Must have extremely pure sample to determine radioactivity per unit mass
- New synthesis avoids water



Conclusions and Future Work

- A 20 min. immersion extracts no statistically detectable amount of RCS
- A small, yet detectable amount of PAG is extracted in under 30 sec.
- A 30 sec. water rinse removes extractable PAG
- Synthesize ^{14}C -triethanolamine
- Perform base extraction experiment



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